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Ecological assessment of Cr (VI) concentrations in the surface waters of Stara Zagora Region used in agriculture

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Abstract. The aim of the present study was ecological monitoring of Cr (VI) concentrations in surface waters from Stara Zagora Region used in agriculture. The research goal was accomplished by: determining Cr (VI) concentrations in real surface water samples taken from 16 sampling points situated in four municipalities: Stara Zagora, Kazanlak, Chirpan and Gurkovo, by AAS (Atomic Absorption Spectrometry) during the period December, 2009 - November, 2010; comparative estimation of the experimental results from the standpoint of national/international surface water quality standards; statistical analyses of the data. The scientific investigations revealed that the II category standard (0.05 mg/L Cr (VI)) was exceeded by 2 % - 30 % in approximately 19 % of the samples analyzed (Tundzha River - bridge Yagoda Village, Radova River, Zetyovo Dam, Chirpan Dam). The registered Cr (VI) concentrations during November, 2010, in 31 % of the surface water bodies practically coincided with the above cited permissible limit.

Keywords: Cr (VI), water bodies, ecological assessment, AAS, Stara Zagora Region

Introduction

The interest in inorganic chromium speciation originates from the widespread industrial use of this metal in the metallurgical, chemical and refractory brick industries. Major applications of hexavalent chromium compounds include metal plating, manufacture of pigments and dyes, corrosion inhibitors, chemical synthesis, refractory production, leather tanning, wood preservation, anodising aluminium in aircraft industries (Giusti and Barakat, 2005; Gowd and Govil, 2008; Vasilatos et al., 2008). Therefore, the quantity of inorganic Cr has been identified as an important water quality index (Yusof et al., 2007). Industrial releases of Cr (VI) compounds to surface water and soil can result in the transport and leaching of these substances into groundwater.

Chromium can exist in oxidation states ranging from -2 to +6, but only +3 and +6 are typically found within the range of pH and redox potential common in environmental systems (Losi et al., 1994; Cieslak-Golonka, 1996; Kotas and Stasicka, 2000; Krishnani and Ayyappan, 2006). Both Cr (III) and Cr (VI) have contrasting biological, geochemical and toxicological effects. Cr (III) is considered to be a trace element species essential for the proper functioning of mammals. It is able to maintain the metabolism of glucose, lipids and proteins, although in excess it can cause allergic skin reactions and cancer (Yusof et al., 2007; El-Sikaily et al., 2007; Miretzky and Cirelli, 2010). Cr (III) has low solubility in water and readily precipitates as Cr(OH)₃ or mixed Cr-Fe hydroxides at pH values greater than 4. The oxidation potential for transforming Cr (III) into Cr (VI) is high, and the probability of transformation into a higher oxidation form in environmental conditions is reduced (Vasilatos et al., 2008).

The environmental impact of Cr (VI) is a controversial issue, critical to the protection of natural water resources (Vasilatos et al., 2008). Cr (VI) is the most toxic form, being carcinogenic and

mutagenic to living organisms. In addition, it leads to liver damage, pulmonary congestion and causes skin irritation resulting in ulcer formation. Inhalation and retention of Cr (VI)-containing materials can cause the increased incidence of bronchogenic carcinoma. Skin contact of Cr (VI) compounds can lead to skin allergies and cancer (Yusof et al., 2007). It is assumed that hexavalent chromium is about 100-1000 times more toxic than trivalent chromium and due to the fact that it has limited hydroxide solubility, it is less mobile and less bioavailable (Yayintas et al., 2007; Miretzky and Cirelli, 2010). Cr (VI) has the ability to permeate biological membranes. Once inside the cell, it is reduced to trivalent chromium, either enzymatically or nonenzymatically - a process called "intracellular" reduction. A series of *in vitro* and *in vivo* studies have demonstrated that Cr (VI) induces an oxidative stress through enhanced production of reactive oxygen species (ROS) leading to genomic DNA damage and oxidative deterioration of lipids and proteins. A cascade of cellular events occur following Cr (VI)-induced oxidative stress including enhanced production of superoxide anion and hydroxyl radicals, increased lipid peroxidation and genomic DNA fragmentation, modulation of intracellular oxidized states, activation of protein kinase C, apoptotic cell death and altered gene expression (Bagchi et al., 2001).

In environmental systems, Cr (VI) exists as oxyanions such as chromate (CrO₄²⁻), bichromate (HCrO₄⁻) and dichromate (Cr₂O₇²⁻) and therefore has high solubility in water and is far more mobile than the trivalent chromium. High Cr (VI) concentrations in surface water bodies are indicators of anthropogenic pollution (Vasilatos et al., 2008) and require the application and implementation of proper treatment methods and/or other quality control strategies, especially when these waters are intended for irrigation or animal watering (Mishra et al., 1995).

The aim of the present study was ecological assessment of Cr (VI) concentrations in surface waters from Stara Zagora Region

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used in agriculture.

Material and methods

A total of 128 surface water samples, taken from 16 sampling points situated in four municipalities, Stara Zagora, Kazanlak, Chirpan and Gurkovo, of Stara Zagora Region, were collected and analyzed during the period December, 2009 - November, 2010 (Table 1). All water samples were collected and preserved according to the standard methods for the examination of water and wastewater (American Public Health Association, 2008). The concentrations of Cr (VI) were determined by atomic absorption spectrometry (AAS) ISO 8288 on AAnalyst 800 (Perkin Elmer) Atomic Absorption Spectrometer. All investigations were performed in triplicate.

The ecological assessment of the surface water quality was done on the basis of national and international surface water quality standards (Regulation No. 7/8.08.1986; USEPA, 1996; WHO, 1997), and statistical analyses of the data.

The statistical significance of the results was tested on the basis of the Standard Deviation (SD) values calculated by the Student's t-test.

Results and discussion

The temporal and spatial trends of natural water quality fluctuations depend on a number of factors such as: physical-geographical conditions, soil content, the hydro-physical, hydro-

chemical and hydro-biological processes occurring in the water bodies (Kostadinova, 2006). The assessment of the anthropogenic impact on the quality of natural water bodies implies the application of the ecosystematic approach (Kostadinova et al., 2006). In the present study, the spatial and temporal variations of the experimentally determined Cr (VI) concentrations in the investigated 128 surface water samples taken from 8 rivers and 3 dams during the period December, 2009 - November, 2010, were traced out.

The analyses of the downstream of Tundzha River quality regarding Cr (VI) contents, including assessment of the influence of its tributaries flowing through the investigated area, is essential and imperative, considering that it is one of the biggest rivers in Bulgaria with about 50 tributaries and the most significant tributary of Maritsa River, flowing into it on the Turkish territory near Edirne, and one of the main water resources for drinking, domestic purposes and irrigation (Barakova et al., 2004). Thus, the concentrations of the toxic metal in four sampling points from the river itself (s.p. 206, 210, 211, 214) and its five tributaries (s.p. 207, 208, 209, 212, 213) passing through the territory of Kazanlak and Gurkovo Municipalities were determined in the present study (Figure 1). The ecological monitoring established elevated Cr (VI) contents in the surface water samples from s.p. 211 (Tundzha River - bridge Yagoda Village) in November 2010 and in s.p. 212 (Radova River) in December 2009 and July 2010. The II category standard for Cr (VI) contents in surface waters, regulated by the Bulgarian legislation, (0.05 mg/L Cr (VI)) in these water bodies was exceeded by 14 % - 30 % (Regulation No. 7/8.08.1986). Besides, the presented concentration curves are characterized with analogous mode - with clearly outlined peak concentrations in s.p. 210, 211 and 212.

To compare and assess the relative influence of the feeders waters on the quality of Tundzha River, Cr (VI) contents was

Table 1. A list of the investigated surface water sampling points in Stara Zagora Region

Municipality	Sampling Point (s.p.)	Code	Coordinates
Stara Zagora	Bedechka River before Zagorka Lake	201	N 42.27049° E 25.37937°
	Bedechka River after Zagorka Lake	202	N 42.32740° E 25.33805°
	Sazliika River - St. Zagora Mineral Baths	204	N 42.26914° E 25.29015°
	Tundzha River (bridge Yagoda Village)	211	N 42.32740° E 25.33805°
	Sazliika River (after Galabovo City) (Gurkovo Municipality)	205	-
Chirpan	Chirpan Dam	215	N 42.09753° E 25.18842°
	Zetjovo Dam	216	N 42.09112° E 25.21541°
	Malak Yurt Dam	217	N 42.19494° E 25.18706°
Kazanlak	Tundzha River (before Koprinka Dam)	206	N 42.41215° E 25.18863°
	Gabrovnitsa River (Dolno Sahrane Village)	207	N 42.38983° E 25.16554°
	Leshnitsa River (after Dunavtsi Village)	208	N 42.38960° E 25.16479°
	Eninska River (after Enina Village)	209	N 42.40004° E 25.24538°
	Tundzha River (after Buzovgrad Village)	210	N 42.34484° E 25.24920°
Gurkovo	Radova River	212	N 42.39392° E 25.46381°
	Lazova River	213	N 42.41709° E 25.47040°
	Tundzha River (Zhrebchevo Dam mouth)	214	N 42.38333° E 25.49350°

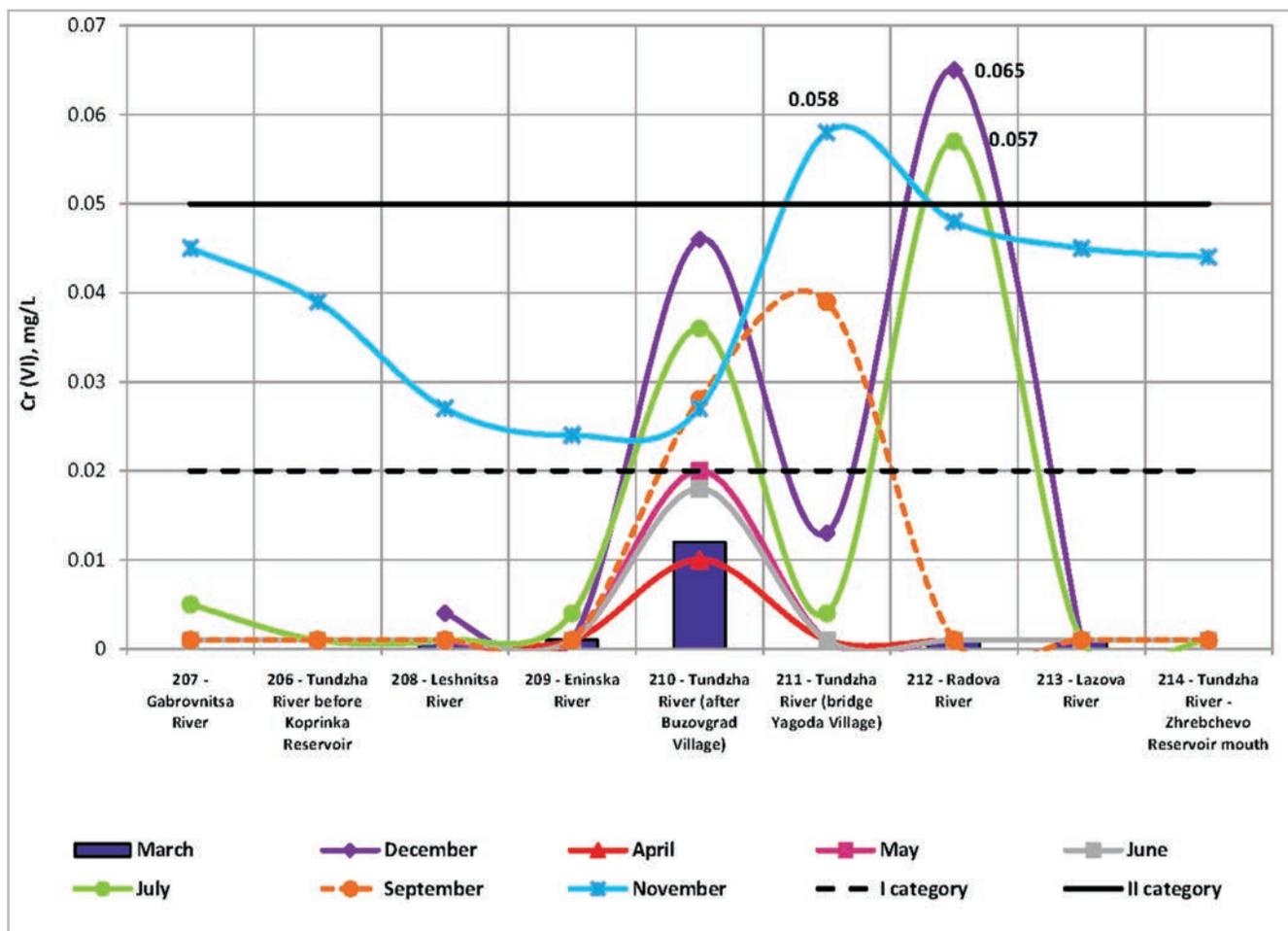


Figure 1. Monthly variations of Cr (VI) concentrations in the surface water samples from the downstream of Tundzha River and five of its feeders

presented in an area plot on monthly bases (Figure 2). It should be outlined that the data displayed was selected to represent logically the measured metal concentrations in the feeder and in the sampling point in the river itself, immediately after its influx. Consequently, the greater Cr (VI) loading of Tundzha River waters in June and September, 2010, could not be attributed to the tributaries waters. However, the registered peaks for both water types in July, 2010, and especially in November, 2010, almost coincided. Thus, the assumption of Cr (VI) mass transfer from the feeders waters into the river seemed logical. Accordingly, these elevated heavy metal concentrations in the upstream and downstream waters of the river could be due to uncontrolled discharge of industrial wastewaters from the wood-processing, machine building and metal plating enterprises situated in the studied area. The latter was sustained by the observations of Kostadinova et al. (2007), which characterized the wastewaters from the town of Kazanlak as irregularly allocated according to their pollution extent, quality indices, temporal and spatial distribution, and established a few pollution cases based on individual quality parameters.

The surface water samples taken from points 201 and 202 (Table 1) were representative for the quality of one of the biggest rivers passing through the territory of Stara Zagora Municipality: Bedechka River - crossing the eastern part of the district center - the city of Stara Zagora, characterized by concentrated industrial activities. The ecological assessment of Bedechka River and Zagorka Lake waters quality, conducted during the period 2004-2005, proved unconfirmities with the permissible standards only

according to the parameters nitrites and nitrates (Kostadinova, 2006). Sampling points 204 and 205 were selected as representatives of the water quality of Sazliika River (Table 1). The measured Cr (VI) concentrations in these surface water bodies are presented in Figure 3. Obviously, the highest levels detected in November, 2010, in s.p. 201, 202 and 204 were only 4 % lower than the II category limit, i.e. practically coinciding with the standard. The mode of the corresponding concentration curves seemed analogous with that observed for Tundzha River (s.p. 211), namely characterized by sharp concentration increase during September and November, 2010.

The experimental concentration curves for Cr (VI) contents in Chirpan (s.p. 215), Zetyovo (s.p. 216) and Malak Yurt (s.p. 217) dams, Chirpan Municipality (Figure 4), also demonstrated elevated heavy metal levels during September and November, 2010. The sharply outlined steep concentration leap from values below the I category standard (0.02 mg/L) to levels exceeding the II category standard (0.05 mg/L) was disturbing (Regulation No. 7/8.08.1986). The highest measured Cr (VI) concentrations during both cited months in s.p. 216 were 0.051 mg/L and 0.052 mg/L, respectively. Regarding the temporal variations of Cr (VI) concentrations in the surface waters of Stara Zagora Region, an analogous outlined tendency of elevated chromium contents during September and November in approximately 60 % of the investigated water samples was observed. The assumption of accidental spills and pollution events that had probably occurred in some of the sampling sites during both months could be accepted as a probable pollution

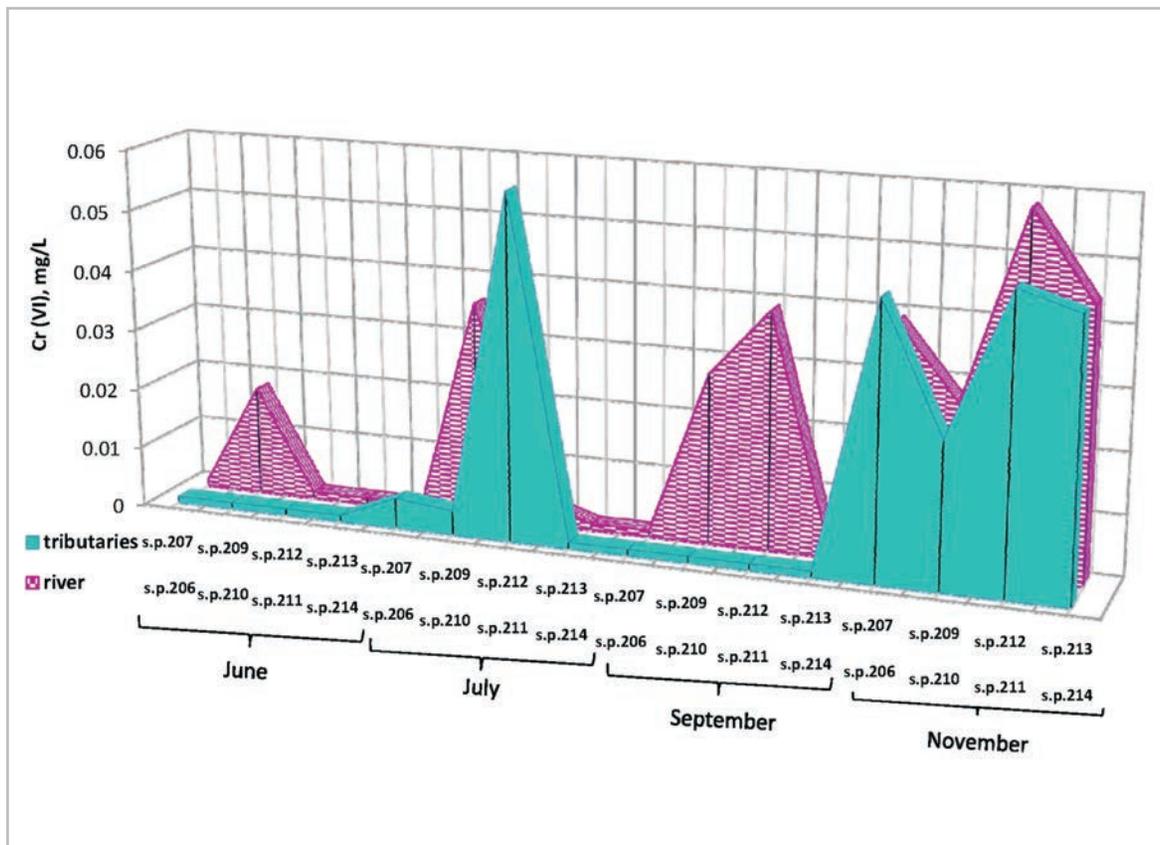


Figure 2. Area plot of Cr (VI) contents in the surface waters of Tundzha River and its feeders during the period June, 2010 - November, 2010

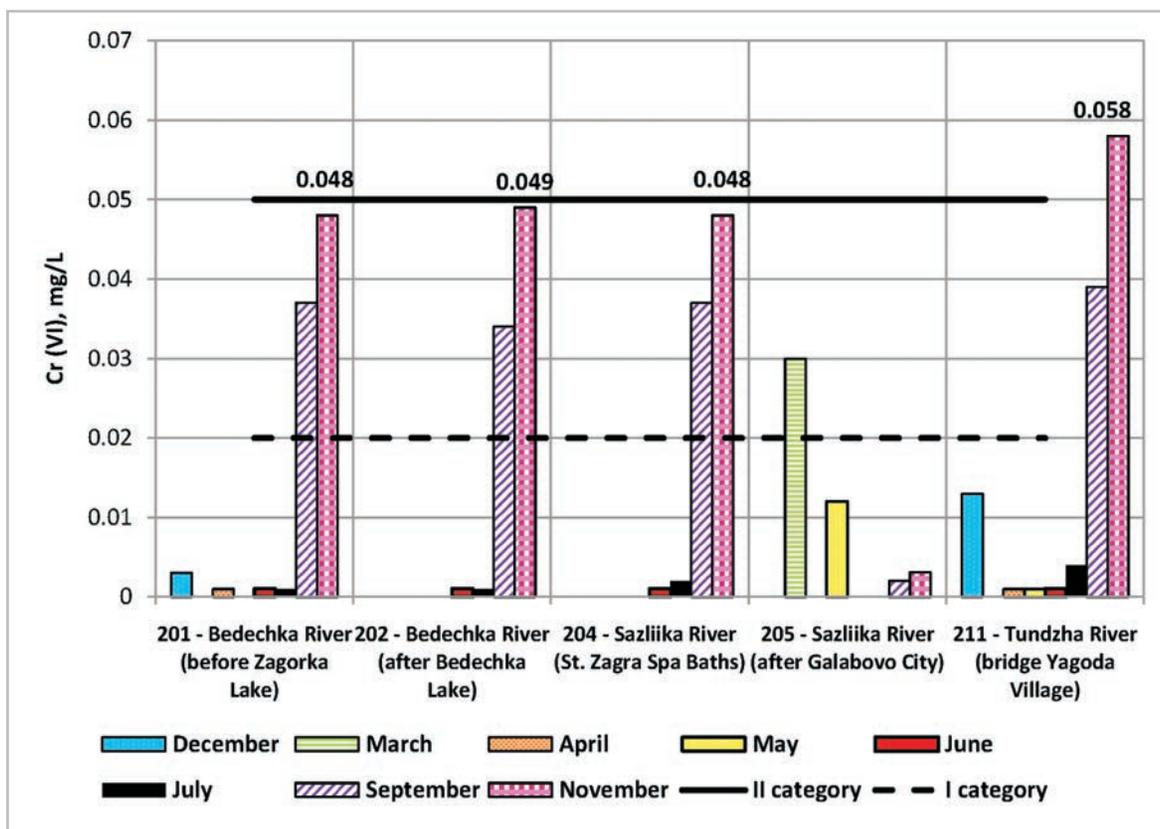


Figure 3. Chromiun (VI) concentrations in the surface waters of Stara Zagora Municipality during the monitored period

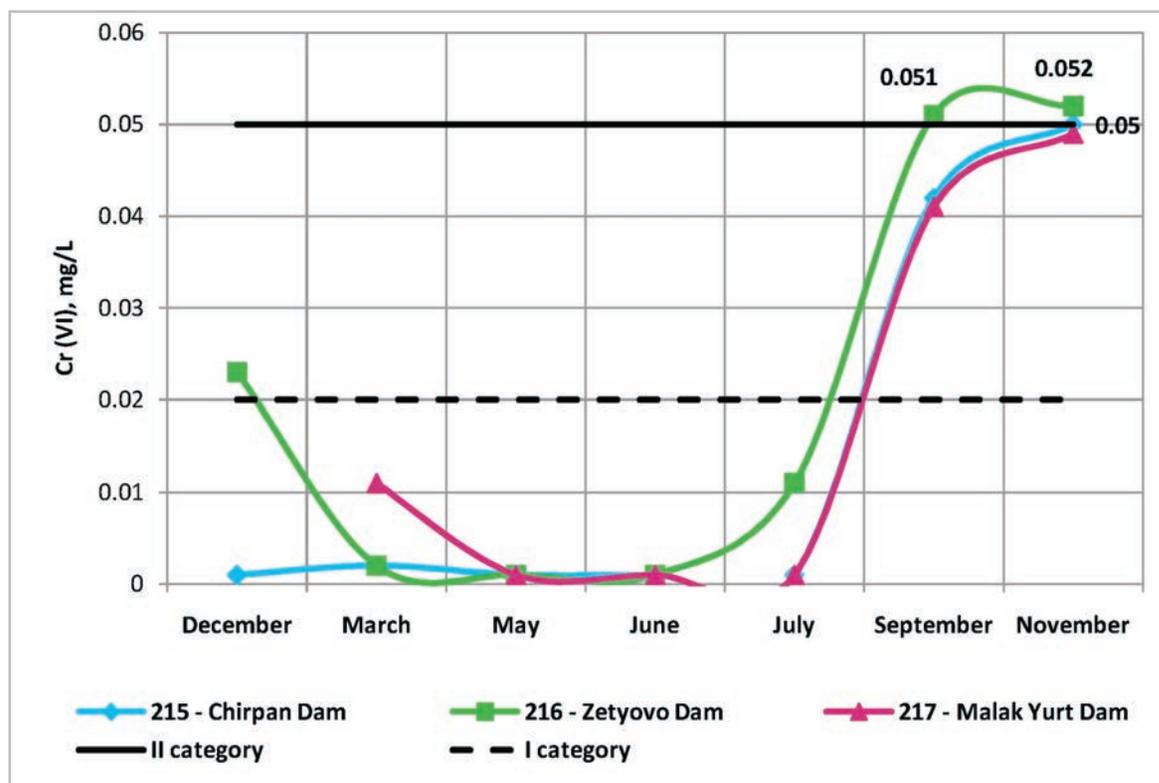


Figure 4. Chromium (VI) concentrations in the dam waters of Chirpan Municipality

Table 2. Maximum, minimum and mean annual Cr (VI) concentrations and SD values

Municipality	s.p.	C*, mg/L	C _{max} , mg/L	C _{min} , mg/L	SD
Stara Zagora	201	0.013	0.048	0.001	0.008
	202	0.021	0.049	0.001	0.012
	204	0.022	0.048	0.001	0.012
	211	0.017	0.058	0.001	0.009
Chirpan	215	0.014	0.050	0.001	0.008
	216	0.020	0.052	0.001	0.009
	217	0.017	0.049	0.001	0.009
Kazanlak	206	0.011	0.039	0.001	0.010
	207	0.013	0.045	0.001	0.011
	208	0.005	0.027	0.001	0.003
	209	0.004	0.024	0.001	0.003
	210	0.030	0.046	0.012	0.007
Gurkovo	212	0.022	0.065	0.001	0.006
	213	0.007	0.045	0.001	0.010
	214	0.012	0.044	0.001	0.011

C* - mean annual concentration; C_{max} - maximum concentration; C_{min} - minimum concentration

resource to some extent. However, the explanation of the observed uniform Cr (VI) concentration curves modes and registered peaks should be sought at a more global scale. Thus, we propose that chromium contents variations in most of the examined surface water samples is mediated by the seasonal variations of the biological activity in the water bodies. Probably, Cr (VI) was reduced to Cr (III) during the periods of high activity, namely the spring and summer months, and the reduced form was removed via adsorption on particles or sediments. Later on, Cr (III) could be released during the remineralisation of the particles and returned in this form to the upper water layers, where it was either oxidised to Cr (VI), or stabilised by the organic compounds, remaining as Cr (III). The latter assumptions were supported by the investigations and observations of Connelly et al. (2006) based on seasonal changes in dissolved chromium speciation in the marine waters of Sargasso Sea.

The values of the maximum (C_{max}), minimum (C_{min}) and mean annual concentrations (C) in all examined surface water bodies measured during the entire monitored period are presented in Table 2. The SD values for all investigated sampling points were calculated and presented in Table 2. The range of their variation, from 0.003 to 0.012, proved the statistical significance of the experimental results. A clear tendency, with regard to the spatial distribution of Cr (VI) in the surface waters of the four municipalities, could not be outlined both because of the commensurable detected heavy metal levels in most of the examined water bodies and the narrow SD range. It is obvious only that the surface waters from Eninska River (s.p. 208) and Tundzha River after Buzovgrad Village (s.p. 209), Kazanlak Municipality are characterized by the lowest C and C_{max} Cr (VI) concentrations and the lowest SD values.

Conclusion

The one-year ecological monitoring of Cr (VI) concentrations revealed that the II category standard (0.05 mg/L Cr (VI)) was exceeded by 2 % - 30 % in approximately 19 % of the samples analyzed: Tundzha River - bridge Yagoda Village, Radova River, Zetyovo Dam, Chirpan Dam. The registered Cr (VI) concentrations in November, 2010, in 31 % of the surface water bodies practically coincided with the above cited permissible limit. The surface waters in Kazanlak Municipality are characterized by the lowest Cr (VI) contents. The generated database could be used to form an integrated Cr (VI) management strategy, as recommended by most up-to-date trends of environmental research in the field of agricultural land use planning aimed at water pollution control.

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Genetics and Breeding	
Selection of oil-bearing rose in Bulgaria – tendencies and perspective	189
N. Kovatcheva	
Combining ability of mutant maize line. I. Number of rows in the ear	193
M. Ilchovska	
Freezing of day 5 and 6 sheep and goat embryos of Greek breeds	196
A. Pampukidou, M. Avdi, R. Ivanova T. Alifakiotis	
Investigation on some seed characteristics among sunflower lines and hybrids	199
M. Drumeva, N. Nenova, E. Penchev	
Determination of coloured horses raised in Turkey	203
O. Yilmaz, M. Ertugrul	
Nutrition and Physiology	
Effects of different levels of dietary digestible amino acids on nitrogen retention and excretion in Topigs pig hybrids	207
A. Ilchev, G. Ganchev	
Development of the caecal microbiota in rabbits weaned at different age	212
B. Bivolarski, G. Beev, S. A. Denev, E. Vachkova, T. Slavov	
Consumption of dissolved oxygen in rainbow trout (<i>Oncorhynchus mykiss</i>)	220
I. Sirakov, Y. Staykov, G. Djanovski	
Effect of coconut oil on rumen and duodenal ammonia concentrations and some blood biochemistry parameters in yearling rams	224
V. Radev, T. Slavov, E. Enev, I. Varlyakov	
Pharmacokinetics of tiamulin and chlortetracycline after application of Tetramutin-premix in pigs	229
D. Dimitrova V. Katsarov, D. Dimitrov, D. Tsoneva	
Production Systems	
Research effect of application of herbicides raft 400 SC for growing of lavender	235
D. Angelova, H. Lambev	
Defining the critical kinematic parameters of rotary harrow with vertical axis of rotation	237
D. Guglev	
Development and experimental study of the maximum temperature potential of a solar thermal module for driving of an absorption air-conditioning machine	240
K. Peychev, R. Georgiev	
Histometrical investigation on the turkey broiler's third eyelid (Harderian) gland	246
D. Dimitrov	
Study of the tolerance of alfalfa varieties (<i>Medicago Sativa</i> L.) to <i>Sitona</i> species (Coleoptera: <i>Curculionidae</i>)	249
I. Nikolova, N. Georgieva	
Productive performance and quality of essential oil from oil bearing rose (<i>Rosa damascena</i> Mill) for use of oxadiargyl	254
D. Angelova	
Study of the thermal efficiency of a solar thermal module at different mounting angles	257
R. Georgiev, K. Peychev	
Behavior of apple rootstock M9 produced by somatic organogenesis in stoolbed	261
G. Dobrevska	
Agriculture and Environment	
Effect of experimentally polluted water on the stomatal and structural characteristics on the leaves of two varieties of <i>Triticum aestivum</i> L. grown on different soil types	265
K. Velichkova, D. Pavlov, D. Ninova	
Ecological assessment of Cr (VI) concentrations in the surface waters of Stara Zagora Region used in agriculture	269
N. Georgieva, Z. Yaneva, D. Dermendzhieva, V. Kotokova	
Effect of shooting on the structure of population of golden jackal (<i>Canis aureus</i> L.) in Sarnena Sredna Gora mountain	276
E. Raichev	
Product Quality and Safety	
Chemical surface disinfection of funnel type fish egg incubators	281
A. Atanasov, N. Rusenova, Y. Staykov, G. Nikolov, A. Pavlov, D. Stratev, E. Raichev	
Fatty acid composition of common carp, rainbow trout and grey mullet fish species	285
M. Stancheva, A. Merdzhanova	

Instruction for authors

Preparation of papers

Papers shall be submitted at the editorial office typed on standard typing pages (A4, 30 lines per page, 62 characters per line). The editors recommend up to 15 pages for full research paper (including abstract references, tables, figures and other appendices)

The manuscript should be structured as follows: Title, Names of authors and affiliation address, Abstract, List of keywords, Introduction, Material and methods, Results, Discussion, Conclusion, Acknowledgements (if any), References, Tables, Figures.

The title needs to be as concise and informative about the nature of research. It should be written with small letter /bold, 14/ without any abbreviations.

Names and affiliation of authors

The names of the authors should be presented from the initials of first names followed by the family names. The complete address and name of the institution should be stated next. The affiliation of authors are designated by different signs. For the author who is going to be corresponding by the editorial board and readers, an E-mail address and telephone number should be presented as footnote on the first page. Corresponding author is indicated with *.

Abstract should be not more than 350 words. It should be clearly stated what new findings have been made in the course of research. Abbreviations and references to authors are inadmissible in the summary. It should be understandable without having read the paper and should be in one paragraph.

Keywords: Up to maximum of 5 keywords should be selected not repeating the title but giving the essence of study.

The introduction must answer the following questions: What is known and what is new on the studied issue? What necessitated the research problem, described in the paper? What is your hypothesis and goal?

Material and methods: The objects of research, organization of experiments, chemical analyses, statistical and other methods and conditions applied for the experiments should be described in detail. A criterion of sufficient information is to be

possible for others to repeat the experiment in order to verify results.

Results are presented in understandable tables and figures, accompanied by the statistical parameters needed for the evaluation. Data from tables and figures should not be repeated in the text.

Tables should be as simple and as few as possible. Each table should have its own explanatory title and to be typed on a separate page. They should be outside the main body of the text and an indication should be given where it should be inserted.

Figures should be sharp with good contrast and rendition. Graphic materials should be preferred. Photographs to be appropriate for printing. Illustrations are supplied in colour as an exception after special agreement with the editorial board and possible payment of extra costs. The figures are to be each in a single file and their location should be given within the text.

Discussion: The objective of this section is to indicate the scientific significance of the study. By comparing the results and conclusions of other scientists the contribution of the study for expanding or modifying existing knowledge is pointed out clearly and convincingly to the reader.

Conclusion: The most important consequences for the science and practice resulting from the conducted research should be summarized in a few sentences. The conclusions shouldn't be numbered and no new paragraphs be used. Contributions are the core of conclusions.

References:

In the text, references should be cited as follows: single author: Sandberg (2002); two authors: Andersson and Georges (2004); more than two authors: Andersson et al.(2003). When several references are cited simultaneously, they should be ranked by chronological order e.g.: (Sandberg, 2002; Andersson et al., 2003; Andersson and Georges, 2004). References are arranged alphabetically by the name of the first author. If an author is cited more than once, first his individual publications are given ranked by year, then come publications with one co-author, two co-authors, etc. The names of authors, article and journal titles in the Cyrillic or alphabet different from Latin, should be transliterated into Latin and article titles should be translated into English. The original language of articles and books translated into English is indicated in

parenthesis after the bibliographic reference (Bulgarian = Bg, Russian = Ru, Serbian = Sr, if in the Cyrillic, Mongolian = Mo, Greek = Gr, Georgian = Geor., Japanese = Ja, Chinese = Ch, Arabic = Ar, etc.)

The following order in the reference list is recommended:

Journal articles: Author(s) surname and initials, year. Title. Full title of the journal, volume, pages. Example:

Simm G, Lewis RM, Grundy B and Dingwall WS, 2002. Responses to selection for lean growth in sheep. *Animal Science*, 74, 39-50

Books: Author(s) surname and initials, year. Title. Edition, name of publisher, place of publication. Example: **Oldenbroek JK**, 1999. *Genebanks and the conservation of farm animal genetic resources*, Second edition. DLO Institute for Animal Science and Health, Netherlands.

Book chapter or conference proceedings: Author(s) surname and initials, year. Title. In: Title of the book or of the proceedings followed by the editor(s), volume, pages. Name of publisher, place of publication. Example:

Mauff G, Pulverer G, Operkuch W, Hummel K and Hidden C, 1995. C3-variants and diverse phenotypes of unconverted and converted C3. In: *Provides of the Biological Fluids* (ed. H. Peters), vol. 22, 143-165, Pergamon Press. Oxford, UK.

Todorov N and Mitev J, 1995. Effect of level of feeding during dry period, and body condition score on reproductive performance in dairy cows, IXth International Conference on Production Diseases in Farm Animals, Sept.11 – 14, Berlin, Germany, p. 302 (Abstr.).

Thesis:

Penkov D, 2008. Estimation of metabolic energy and true digestibility of amino acids of some feeds in experiments with muscovy duck (*Carina moschata*, L). Thesis for DSc. Agrarian University, Plovdiv, 314 pp.

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